DITERPENES AND OTHER CONSTITUENTS FROM AUSTRALIAN HELICHRYSUM AND RELATED SPECIES

J. JAKUPOVIC, A. SCHUSTER, F. BOHLMANN, U. GANZER, R. M. KING* and H. ROBINSON*

Institute of Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, F.R.G.; *Smithsonian Institution, Dept. of Botany, Washington D.C. 20560, U.S.A.

(Received 6 June 1988)

Key Word Index—Helichrysum ambiguum, H. bilobum, H. davenportii, H. leucopsideum, H. lindleyi; Craspedia glauca, C. pleiocephala; Bellida graminea; Compositae; diterpenes; ent-beyerenes; ent-labdanes; seco-diterpenes; sesquiterpenes; azulenes; cadalene; guaiane, new skeleton.

Abstract—The investigation of several Australian representatives of the genus *Helichrysum* and related genera afforded 28 new diterpenes (20 ent-labdanes, five seco ent-labdanes, one iso-kaurene, two pimarenes), a cadalenal, a guaiane and a sesquiterpene alcohol with a new carbon skeleton.

INTRODUCTION

The taxonomy of the Australian Helichrysum and related species is difficult and the generic limits are not clear in all cases [1]. In addition, a separation of the Australian Helichrysum species from the South African species has been proposed [1]. In continuation of our chemical investigations of members of the Australian representatives of the subtribe Gnaphaliinae (Compositae, tribe Inuleae) [2–9], we have now studied five Helichrysum, two Craspedia and a Bellida species. The results are discussed in this paper.

RESULTS AND DISCUSSION

The aerial parts of Helichrysum ambiguum Turcz. subsp. ambiguum afforded, in addition to widespread compounds (Experimental), a complex mixture of diterpenes from which we obtained the ent-labdanes 1-19, the seco derivatives 20-24, the sesquiterpene aldehyde 25 and the known azulenes 26-28 [3, 6, 10]. Comparison of the ¹H NMR spectra of 1–10 (Table 1) with those of known compounds showed that we were dealing with manoyloxide derivatives with three oxygen functions, one always being a keto group. In the case of the diol 1, spin decoupling showed that the keto group was at C-7 while the two hydroxy groups had to be placed at C-1 and C-3. The observed couplings of H-1 and H-3 indicated an axial orientation of these groups. In the spectrum of 2, one of these carbons now carried an acetoxy group. The relative position could be deduced by the observed NOE's which also established the proposed stereochemistry at all chiral centres. Thus irradiation of H-16 gave NOE's with H-14 (6%), H-15t (3%) and H-17 (8%), H-17 showed effects with H-6a (8%), H-16 (4%) and H-20 (10%), H-18 with H-3 (8%), H-5 (7%), H-6 β (9%) and H-19 (4%), H-19 with H-2 α (5%), H-3 (8%), H-6 α (4%) and H-20 (5%) while H-20 gave NOE's with H-1 (5%), H-2 β (12%), H-6 α (8%), H-17 (9%) and H-19 (6%). As the signal of H-3 was shifted downfield in the spectrum of 2 compared to that in 1 a 3β -acetoxy derivative was present. The $^{13}CNMR$ data of 1 and 2 (Table 2) also supported the structures. A positive Cotton effect at 293 nm required an *ent*-labdane. The 1H NMR spectrum of 3 showed that it was the 1β -acetoxy isomer of 2.

The ¹H NMR spectrum of 4 and the corresponding acetate, obtained by acetylation of the natural carbinol, showed that they were 3β -angeloyloxy derivatives. Accordingly, the chemical shifts of 4 were similar to those of 2 but differed from those of 3. The acetate of 4 was not identical with 8 (see below) which indirectly established the position of the acetate group in the latter. Identical couplings indicated the same stereochemistry. The ¹H NMR signals in the spectrum of 5 were similar to those of 3, therefore an 1β -angeloyloxy derivative was present.

The ¹H NMR spectra of 6–10 (Table 1) showed that we were dealing with diesters of the diol 1. The nature of the ester groups could be deduced from the typical signals of the ester residues. The relative positions of the different ester groups were arrived at by comparing the chemical shifts of H-1 and H-3 as well of those of the neighbouring protons including those of the mixed diesters 13–16 (see below) were in one case the relative position could be determined by the observed NOE's. However, in a few cases the assignment may be doubtful.

Inspection of the ¹H NMR spectra of 11–16 (Table 3) and spin decoupling showed that these ketones had an additional oxygen function at C-11. The chemical shift and the couplings of H-11 required an axial hydroxy group. NOE difference spectroscopy with the angelate 12 showed that the stereochemistry was the same as in the ketones 1-10. Thus irradiation of H-16 gave effects with $H-12\alpha$ (6%), H-14 (6%), H-15t (3%) and H-17 (6%), of $H-12\alpha$ 17 with H-6 α (5%), H-16 (3%) and H-20 (7%), of H-18 with H-3 (8%), H-5 (8%) and H-6 β (10%), of H-19 with $H-2\alpha$ (4%), H-3 (9%), $H-6\alpha$ (5%), and H-20 (10%) and of H-20 with H-1 α (7%), H-2 α (3%), H-6 α (6%), H-17 (11%) and H-19 (6%). The nature of the ester groups again followed from the typical signals and in the case of the diester 16 the relative position of the ester groups could be determined by the observed NOE's. In addition to the

1 2 3 4 4Ac 5 10 11 12 13 14 15 16 6 H Ac H Ang Ang H Epang Epang Ac Ac Ang H Ang Epang Epang Αc Ang R² H H Ac H Ac Ang Ang Sen Ang Sen iVal H Ang Ac Epang Tigl R³ H H H H н н Н Н Н Н н он он он OH OHOH

24

29a

R CO₂Me

26 R = Me 27 R = CHO 28 R = CO₂Me

same effects as in the case of 2, irradiation of H-18 gave an effect with the methyl group (H-5') of the epoxyangelate due to the presence of a 3β -epoxyangeloyloxy group. The 13 C NMR spectra of 12 and 16 supported the proposed structures (Table 2).

The ¹H NMR spectrum of 18 (Table 4) was in part similar to that of 4. However, the chemical shifts of the double doublets at δ 2.42, 2.69 and 3.15 indicated a second keto group. The assignment of these signals was achieved by the observed NOE's. Thus irradiation of H-17 showed an effect with H-11 α (4%) and also with H-6 α (4%) and H-20 (9%). Further effects were observed between H-18, H-3 (6%), H-5 (7%), H-6 β (8%), H-19 (3%) and the methyl signal of the angelate (H-5') which established the relative position of the ester group. The remaining NOE's were identical with those of compound 2. The diketone 18 showed a positive Cotton effect. The ¹³C NMR spectrum (Table 2) also agreed with the structure. The ¹H NMR spectrum of 17 (Table 4) showed that the corresponding 3β -acetoxy derivative was present while that of 19 (Table 4) was in part similar to that of 18, however, an additional low field signal at δ 3.76 and threefold doublets at δ 2.04 and 1.82 showed the presence of an axial 12-hydroxy group. The couplings indicated an equatorial orientation of the latter.

The ¹H NMR spectra of **20** and **21** (Table 5) showed that these diterpenes only differed in the nature of the ester groups, **20** being an acetate and **21** an angelate. Though several parts of the spectra resembled those of **2**

Table 1. ¹H NMR spectral data of compounds 1-10 and 4Ac (400 MHz, CDCl₃, δ-values)

H	-	2	3	4	4Ac	\$	9	7	œ	6	01	Multi- plicity
_	3.59	3.61	4.91	3.62	4.73	5.03	4.90	4.88		4.75	4.75	pp
2α	2.20	2.31	2.26	2.36	2.24	2.31	2.30	2.34		2.27	2.23	ppp
2β	1.98	1.91	1.99	1.96	2.14	2.01	2.03	1.97		2.08	2.02	ppp
· 60	3.62	4.97	3.43	5.06	4.73	3.43	4.80	4.73		4.74	4.74	qq
2	2.12	2.04	2.11	2.04	2.28	2.12	2.30	2.30		2.28	2.29	pp
2α	2.62	2.65	2.65	2.66	5.66	2.66	2.67	5.66		2.64	2.65	рp
6β	2.34	2.38	2.43	2.40	2.4	2.43	2.45	2.43		2.40	2.41	рp
6	2.38	2.34	2.23	2.37	2.27	2.16 dd	2.23	2.23		2.22	2.26	W.
11	-97	(1.65-	(1.8–1.6	} }	1.70	1.68	1.65	1.65		1.65	16	u
	<u> </u>	}			1.31	1.36	(1.35	1.35		1.32	1.8	ш
12	(1.8	(1.8	(1.34	1.8	\$. 2	1.79	2.1. 2.2.	ς ς	5.7	5.7	1 30	E 8
4	5.90	5.93	5.93	5.95	5.94	5.86	5.87	5.92		5.90	5.92	pp
<u>ς</u>	4.93	4.96	4.96	4.96	4.96	4.86	4.89	4.91		4.88	4.98	qq
₹	5.18	5.22	5.20	5.21	5.20	5.15	5.19	5.15		5.13	5.24	qq
9	1.30	1.33	1.30	1.34	1.32	1.29	1.32	1.32		1.28	1.32	S ₃
7	1.48	1.51	1.49	1.52	1.51	1.50	1.49	1.52		1.49	1.53	s
œ	0.84	0.85	88.0	0.87	0.93	68'0	0.93	0.92		98.0	0.87	S
6	0.95	0.93	1.00	96:0	86.0	1.00	1.00	0.98		0.93	0.95	s
0	0.97	1.01	1.09	1.03	1.11	1.11	1.13	1.11		1.08	1.10	s
~		2.07 s	2.07 s	6.12 44	1.91 s	6.14 qq	6.19 <i>qq</i>	5.64 44	s	1.92 s	2.01 s	
		2.62 d(OH)	_	1.99 dq	6.08 44	2.01 dq	2.07 dq	2.19 d	Ьb	5.62 qq	2.1-	
				1.87 dq	2.00 dq	1.87 dq	1.91 dq	1.90 d	фd	2.16 d	2.2 m	
				2.70 d(OH)	1.89 dq	2.33 brs(OH)	2.91 q	2.82 q	dq	1.88 d	0.94 d	
							1.53 s	1.52 s				
								~				

J[Hz]: $1.2\alpha = 1.2\beta = 2\alpha.3 = 2\beta.3 = 5.6\beta \sim 3$; $2\alpha.2\beta = 16$; $5.6\alpha = 6\alpha.6\beta = 14.5$; $9.11\beta = 4.5$; 14.15c = 10.8; 14.15c = 17.5; 15c.15t = 1; compound 2: 1, OH = 10; compound 4: 1, OH = 9; OAng: 3.4 = 7, 3.5 = 4.5 = 1.5; OEpang: 3.4 = 5.5; OSen: 2.4 = 2.5 = 1.2; OiVal: 3.4 = 3.5 = 7.

Table 2. ¹³C NMR spectral data of compounds 1, 2, 12, 16, 18, 20 and 22 (100 MHz, CDCl₃)

C	1	2*.+	12	16		18	20*	22	Multiplicity
1	71.9	71.0	74.2	72.4		70.3	80.6	80.4	d
2	29.4	28.4	28.3	26.4		28.3	28.9	24.6	t
3	77.4	78.1	75.4	76.5		77.0	76.1	76.0	d
4	38.2	37.2	38.3	36.9		37.3	38.3	37.6	S
5	43.0	43.9	43.8	44.8		43.9	37.2	41.8	d
6	35.1	34.8	35.4	35.3		35.5	33.0	31.7	t
7	209.1	208.6	208.4	208.3		209.1	178.6	178.0	S
8	80.7	80.6	80.3	80.3		80.3	106.0	105.5	S
9	47.7	48.4	48.3	48.3		45.3	58.2	55.3	d
10	41.4	41.2	41.2	41.1		41.1	46.4	47.6	S
11	14.6	14.6	65.2 d	65.3 d		37.3	19.5	19.1	t
12	34.3	34.6	43.8	44.0		$207.0 \ s$	29.9	31.3	t
13	74.5	74.5	73.5	73.5		83.5	73.7	73.4	S
14	146.5	146.4	146.0	146.1		140.1	147.7	144.4	d
15	111.2	111.1	112.1	111.9		115.5	111.0	111.6	t
16	29.0	28.6	30.6	30.4		27.2	29.1	29.1	q
17	24.3	24.1	26.3	26.2		21.3	29.9	31.5	q
18	27.7	26.9	28.2	27.8		26.3	25.9	27.1	q
19	20.9	20.7	21.6	21.6		21.0	22.7	21.9	q
20	15.3	15.5	16.6	16.6	,	14.8	25.0	14.6	q
OR		169.3 s	166.3 s	168.0 s	166.6 s	166.0 s	171.1 s	170.0	S
		21.1 q	126.7 s	59.7 s	128.3 s	126.7 s	21.3 q	21.3	q
		•	141.0 d	59.6 d	138.8 d	140.3 d			
			15.6 q	19.0 q	14.5 q	$15.8 \ q$			
			20.6 q	13.3 a	12.0 q	20.6 q			

Assignment with aid of *2D-hetero correlated and †selective INEPT-experiment.

Table 3. 1 H NMR spectral data of compounds 11–16 (400 MHz, CDCl₃, δ -values)

Н	11	12	13	14	15	16	Multiplicity
1	3.95	5.38	4.79	4.90	4.90	4.90	dd
2α	2.46	2.39	2.34	2.43	2.35	2.40	ddd
2β	2.00	2.05	2.16	2.09	2.02	2.04	ddd
3	5.09	3.44 br s	5.14	5.16	5.05	5.15	dd
5	2.01	2.04	2.20	2.22	2.22	2.24	dd
6α	2.75	2.75	2.75	2.76	2.76	2.76	dd
6β	2.43	2.46	2.45	2.47	2.45	2.48	dd
9	2.33	2.13	2.15	2.14	2.25	2.21	d
11	4.67	4.08	4.13	4.08	4.12	4.05	ddd
12α	1.88 m	1.87	1.89 m	$\{1.88 \ m$	1.97	1.92	dd
12β (1.00 m	1.98	1.09 m	1.00 m	1.84	1.84	dd
14	5.91	5.82	5.84	5.85	5.90	5.82	dd
15c	4.94	4.87	4.90	4.90	4.92	4.81	dd
15t	5.18	5.14	5.19	5.20	5.19	5.15	dd
16	1.43	1.42	1.47	1.47	1.47	1.43	S
17	1.81	1.78	1.81	1.82	1.81	1.81	S
18	0.87	0.99	0.88	0.92	0.93	0.93	S
19	0.99	0.91	0.99	1.01	1.01	1.02	S
20	1.51	1.51	1.52	1.54	1.52	1.53	S
OR	6.13 qq	6.14 qq	6.14 qq	6.20 qq	5.61 qq	6.89	qq
	2.00 dq	2.02 dq	2.03 dq	2.07 dq	2.20 d	1.82	br s
	1.86 dq	1.83 dq	1.87 dq	1.88 dq	1.90 d	1.81	br d
			$1.91 \ s$	2.90 q	2.80 q	2.80	q
				1.46 s	1.51 s	1.44	S
				1.14 d	1.18 d	1.13	d

Couplings as in Table 1 except 9,11 = 4.5; $11,12\alpha = 11,12\beta = 6.5$; OTigl: 3,4 = 7.

Table 4. ¹H NMR spectral data of compounds 17–19 (400 MHz, CDCl₃, δ -values)

Н	17	18	19
1	3.45 ddd	3.46 ddd	3.65 br s
2α	2.30 ddd	2.35 ddd	2.35 ddd
2β	1.94 ddd	1.97 ddd	1.99 ddd
3	5.01 dd	5.09 dd	5.01 dd
5	2.10 dd	2.10 dd	2.17 dd
6α	2.63 dd	2.63 dd	2.67 dd
6β	2.51 dd	2.52 dd	2.42 dd
9	3.12 dd	3.15 dd	2.83 dd
11α	2.42 dd	2.42 dd	2.04 ddd
11 <i>B</i>	2.70 dd	2.69 dd	1.82 ddd
12	_	_	3.76 dd
14	5.81 d	5.80 d	5.89 dd
15c	5.19	5.17 dd	5.26 dd
15t	5.43	5.40 dd	5.56 dd
16	1.45	1.44 s	1.52 s
17	1.37	1.36 s	1.39 s
18	0.91	0.93 s	0.89 s
19	0.97	0.99 s	$0.97 \ s$
20	1.03	1.04 s	1.04 s
OR	2.68 d (OH)	2.74 d (OH)	6.07 qq
	2.12 s	6.14 qq	1.97 dq
		1.99 dq	1.88 dq
		1.88 dq	

Couplings as in Table 1 except compounds 17 and 18: 1,OH=9; $9,11\alpha=12.5$; $9,11\beta=7$; $11\alpha,11\beta=19$; compound 19: $9,11\alpha=12$; $9,11\beta=3$; $11\alpha,11\beta=13.5$; $11\alpha,12=4$; $11\beta,12=4.5$.

and 4, the changed chemical shifts of H-5 and H-6 and the 13 C NMR spectrum of **20** (Table 2) clearly indicated that an acid was present (C-7: δ 178.6). Furthermore, a singlet at δ 106.0 required an acetalic carbon. In addition to the results of spin decouplings, the observed NOE's led to the structure **20**. Thus irradiation of H-20 gave NOE's with H-1 (7%), H-6 (2%), H-6' (4%), H-9 (6%), H-17 (3%) and H-19 (4%), of H-18 with H-3 (5%), H-5 (6%), H-6 (2%), H-19 (4%) and the acetate methyl (2%), of H-17 with H-1 (3%), H-9 (6%), H-14 (4%) and H-20 (3%), of H-16 with H-12 (5%) and H-14 (6%), of H-5 with H-11 (3%) and H-18 (4%) as well as of H-1 with H-17 (3%), H-19 (4%) and H-20 (6%). These results required the proposed structure and stereochemistry.

The ¹H NMR spectra of 22 and 23 (Table 5) showed that again these diterpenes only differed in the nature of the ester groups. Comparison of the chemical shifts in the spectra of 20 and 22 required differences in the stereochemistry. In particular, the shifts of H-9, H-14 and H-20 showed pronounced differences. The observed NOE's clearly indicated that we were dealing with 8,9-bis-epimers. Thus H-16 gave NOE's with H-14 (5%) and H-17 (6%), H-17 with H-5 (5%), H-9 (5%) and H-16 (7%) while H-20 gave only effects with H-1 (4%), H-6' (3%), H-11 (4%) and H-19 (3%) but not with H-9 and H-17. The ¹³C NMR data of 22 (Table 2) also supported the proposed stereochemistry.

The ¹H NMR spectrum of **24** (Table 5) was similar to that of **4**. However, the molecular formula indicated an additional oxygen, most likely at C-8 as was deduced from the downfield shift of H-17. Inspection of a model

showed that the observed couplings agreed nicely with those which should be present in the proposed stereochemistry of 24. Obviously, 24 is the precursor of 21 and 23 which may be formed by ring opening of 24 to an acid with an enol ether bond (Δ^8) followed by addition of the 1-hydroxy group either from the α - or the β -face.

The ¹H NMR spectrum of **25** (Experimental) indicated the presence of a derivative of cadalene. A low field singlet at δ 10.31 required an aldehyde group. Spin decoupling allowed the assignment of all signals. The resulting sequences only agree with a cadalene with an aldehyde group at C-9.

The aerial parts of *H. bilobum* Waluf, subsp. *bilobum* afforded, in addition to widespread compounds, the sesquiterpene 30, while the roots gave the sesquiterpene 29 and the diterpene 31. The structure of the latter followed from its ¹H NMR spectrum (Experimental) which was similar to that of the corresponding diacetate [11]. The relative position of the ester groups followed from the observed NOE between H-17 and H-2 of the senecioate residue.

The structure of 30 followed from the 1H NMR spectrum (Experimental). The presence of a methyl ester of an α -substituted acrylic acid could be deduced from the characteristic signals and spin decoupling allowed the assignment of all signals though several were multiplets. Reaction with potassium hydroxide gave guaia-4-11(13)-dien-12,8 β -olide [12]. The 13 C NMR data supported the structure of 30.

The ¹H NMR spectrum of 29, its ¹³C NMR data, spin decoupling and NOE difference spectroscopy indicated the presence of an unusual sesquiterpene alcohol with an eight-membered ring. Only at elevated temperature in deuteriobenzene was a clear ¹H NMR spectrum obtained. Spin decoupling allowed the assignment of all signals and the stereochemistry was determined by the observed NOE's. Clear effects were obtained between H-14 and H-5 (10%), between H-12 and H-5 (6%), between OH, H-11 (5%) and H-13 (5%), between H-12, H-8 (6%) and H-13 (5%) as well as between H-13 and H-6 β (6%). The ¹³CNMR data (Experimental) also supported the proposed structure. Most likely the carbinol 29 is formed by cyclization of 4Z-α-humulene (Scheme). We have named the carbon skeleton ambiguane which so far has only been prepared by cyclication of zerumbone epoxide [13].

The aerial parts of H. davenportii F. Muell. afforded isokaurenic acid 33 and the thiophene derivative 37. The structure of the latter was deduced from the spectral data which were similar to those of the corresponding desoxy derivative 37a [14] but different from those of the isomeric alcohol with a hydroxypropenyl side chain [14]. The structure of 33, molecular formula C₂₀H₃₀O₃, followed from the spectral data. In deuteriobenzene all signals in the ¹H NMR spectrum (Experimental) could be assigned by spin decoupling and by the observed NOE's which also established the stereochemistry. Clear effects were observed on irradiation of H-20 with H-6a (8%), H- 2α (7%) and H-11 (6%), of H-17 with H-12 (5%) and H-15 (6%), of H-13 with H-12 (10%), of H-12 with H-13 (10%), H-17 (6%) and H-11 β (6%), of H-18 with H-5 (8%), H-3 α (6%) and H-6 β (7%), of H-9 with H-15 (4%) and of H-1 β with H-9 (7%). The ¹³C NMR data also supported the structure.

The aerial parts of *H. lindleyi* H. Eichler only gave the thiophenes 37 and 38 and triterpenes while those of *H. leucopsideum* DC gave some flavones.

548 J. Jakupovic et al.

Table 5.	¹ H NMR	spectral	data	of	compounds	20-24	and	22Me	(400 MHz,
			Cl	DC.	l_3 , δ -values)				

Н	20	21	22	22Me	23	24	Multiplicity
1	3.73	3.69	3.96	3.95	3.97	3.75	dd
2α	2.13	2.20	2.35	2.34	2.31	2.33	ddd
2β∫	2.13	2.12	1.89	1.89	1.89	1.98	ddd
3	4.63	4.78	4.55	4.53	4.74	5.07	dd
5	2.78	2.82	2.50	2.49	2.52	1.98	dd
6	2.45	2.45	2.43	2.39	2.42	2.77	dd
6′	2.34	2.36	2.18	2.11	2.17	2.60	dd
9	1.70	1.68	1.96	1.97	1.95	2.73	dd
11))	1.7	1.36	1.35	1.36	1.85	m
11'	1.7-	1.7	1.30	1.28	1.30	1.27	m
12		1.9	1.65	1.63	1.66	2.25	m
ر'12	1.60	1.56	1.45	1.44	1.45	1.75	m
14	6.02	6.05	5.66	5.65	5.65	6.26	dd
15c	4.97	4.95	4.96	4.99	4.99	4.99	dd
15t	5.07	5.05	5.26	5.26	5.23	5.10	dd
16	1.30	1.32	1.31	1.31	1.30	1.26	S
17	1.35	1.24	1.65	1.68	1.62	1.73	S
18	0.88	0.88	0.87	0.80	0.86	0.90	S
19	0.99	1.02	1.00	0.99	1.03	1.02	S
20	1.10	1.10	0.98	0.88	0.90	0.99	S
OR	2.08 s	6.07 qq	2.03 s	2.02 s	6.02 qq	6.15	qq
		2.04 dq		$3.70 \ s$	2.01 dq	2.00	dq
		1.97 da			$1.91 \ dq$	1.88	dq

J[Hz]: 14,15c=10.5; 14.15t=17.5; 15c,15t=1; compounds **20** and **21**: 1,2 α = 1,2 β =6; 2 α ,2 β =16; 2 α ,3=2 β ,3=5; 5,6=9; 5,6'=2; 6,6'=16.5; 9,11=7: 9,11'=10.5; compounds **22** and **23**: 1,2 α =1,2 β =2 α ,3=2 β ,3=3; 2 α ,2 β =16; 5,6=8; 5,6'=1.5; 6,6'=16; 9,11=8; 9,11'=11; compound **24**: 1,2 α =1,2 β =2 α ,3=2 β ,3=3; 2 α ,2 β =16; 5,6=11; 6,6'=15; 9,11=2; 9,11'=12.

The aerial parts of Craspedia pleiocephala F. Muell. afforded the pimarene derivative 32 which was isolated as its methyl ester. The structure followed from its 1 H NMR spectrum (Experimental) which was similar to that of related diterpenes. The nature of the ester group followed from the 1 H NMR signals and the fragmentation pattern in the MS. The stereochemistry was determined by NOE difference spectroscopy. Clear effects were observed on irradiation of H-20 with H-1 α (5%), H-11 α (5%) and H-19 (10%), of H-18 with H-5 (8%), H-6 α (7%) and H-15 (8%), of H-19 with H-14 α (8%), H-16 (7%) and H-15 (8%), of H-19 with H-6 (6%) and of H-7 with H-14 α (10%) and H-6 β (7%).

The aerial parts of *Craspedia glauca* (Labill.) Sprengel gave, in addition to widespread compounds (Experimental), the *ent*-beyerene derivatives **34–36**.

From the aerial parts of *Bellida graminea* Ewart also thiophene acetylenes were isolated. In addition to **37a** and **38**, 2-but-3-en-1-ynyl dithiophene was present as well as luteolin and quercetin, while the aerial parts of *Millotia myosotidifolia* (Benth.) Steetz. gave no characteristic compounds.

The chemistry of the Australian Helichrysum species differs from that of the South African ones, especially, by the accumulation of diterpenes not very common in the latter. Furthermore, these constituents allowed a clear separation from the Australian Helipterum species which can be characterized by the presence of sesquiterpene lactones [5]. The presence of an ent-beyerene in Craspedia shows relationships to Myriocephalus [8], while the

thiophene acetylenes in *Bellida* are not very characteristic as these compounds are reported from many representatives of the whole tribe and many other Compositae. Further investigations may show whether the chemistry is helpful in resolving the complicated taxonomy of the Australian Gnaphaliinae.

EXPERIMENTAL

The air-dried plant material was collected in August 1986 in Australia, vouchers are deposited in the US National Herba-Washington. Extraction was achieved with MeOH-Et₂O-petrol (1:1:1) and the extracts were worked-up and separated as reported previously [16]. Final separation conditions for new compounds are placed together with the spectral data (TLC: Tl CH₂Cl₂-Et₂O, 2:1; T2 CH₂Cl₂-Et₂O, 9:1; T3 Et₂O-petrol, 3:1; T4 Et₂O-petrol, 2:1; T5 Et₂O-petrol, 1:1; HPLC (RP 8, ca 100 bar, flow rate, 3 ml/min): HPl MeOH-H₂O, 3:2; HP2 7:3; HP3 17:3). The extract of the aerial parts of Helichrysum ambiguum (550 g) (voucher RMK 9630) gave by CC four fractions (1: petrol; 2: Et₂O-petrol, 1:3; 3: Et₂O-petrol, 1:1 and 4: Et₂O and Et₂O-MeOH, 9:1). Fraction 1 gave by TLC 10 mg caryophyllene, 25 mg bicyclogermacrene and 25 mg γ-humulene. Fraction 2 afforded ca 7 mg each of 26-28, 7 mg spathulenol, 3 mg 25 (TLC, Et₂O-petrol, 1:9, three developments, R_{1} 0.41) and 10 mg caryophyllenepoxide. Fraction 3 contained a mixture of triterpenes and fraction 4 was further separated by medium pressure CC. Finally by TLC and HPLC (conditions see Table 6) 10 mg 21, 10 mg 23, 70 mg 4, 20 mg 20, 20 mg 22, 10 mg 5, 6 mg 8 and 9 (ca 1:1), 2 mg 10, 3 mg

13, 2 mg 24, 150 mg 2, 5 mg 12, 3 mg 18, 15 mg 6 and 7 (ca 2:1), 3 mg 11, 7 mg 14, 2 mg 15, 5 mg 16, 50 mg 1, 2 mg 17, 70 mg 3 and 10 mg 19 were obtained (raising polarity). Data of 1-24: Table 6.

38

Helichrysum bilobum (voucher RMK 9605, 500 g aerial parts). CC and TLC gave 50 mg bicyclogermacrene, 5 mg α - and 5 mg γ -gurjunene, 50 mg caryophyllene, 150 mg of its epoxide, 300 mg spathulenol, 10 mg ent-kaurenic acid, 10 mg betulin and 470 mg 30 (T5 R_f 0.77). The roots (260 g) gave 10 mg 29 (T4 R_f 0.51), 2 mg 31 (T4 four developments, R_f 0.85), 10 mg 5-hydroxyobliquin and 20 mg stigmasterol.

Helichrysum davenportii (voucher RMK 9585, 500 g aerial parts). CC, TLC and HPLC afforded 4 mg 37 (HP3 R, 2.7 min) and 20 mg 33 (HP3 R, 4.6 min).

Helichrysum lindleyi (voucher RMK 9526, 50 g aerial parts). CC and TLC gave 2 mg 37, 3 mg 38, 3 mg stigmasterol and 2 mg sitosterol.

Helichrysum leucopsideum (voucher RMK 9601, 470 g aerial parts). CC and TLC gave a complex mixture (100 mg) of flavones.

Craspedia glauca (voucher RMK 9606, 300 g aerial parts). CC and TLC gave 3 mg β -selinene, 20 mg costal, 5 mg ent-kaurenic acid, 5 mg trachylobanic acid, 15 mg 34, 20 mg 35, 10 mg 36 and 10 mg isoalantolactone.

Craspedia pleiocephala (voucher RMK 9616, 260 g aerial

parts). TLC of the polar fractions gave 5 mg 32, isolated as its methyl ester 32a (T5 R_f 0.62).

Bellida graminea (voucher RMK 9589, 100 g aerial parts). CC and TLC gave 45 mg bisabola-1,3(15),9-triene, 30 mg 37, 65 mg 2-but-3-ent-1-inyl dithiophene, 25 mg 37a, 60 mg quercetin and 50 mg luteolin.

Millotia myosotidifolia (voucher RMK 9623, 230 g aerial parts) gave no characteristic compounds.

Cadalen-14-al (25). Colourless oil; IR $v_{\text{max}}^{\text{CC1}_{k}}$ cm $^{-1}$: 2740, 1690 (CHO); MS m/z (rel. int.): 212.120 [M] + (100) (calc. for C_{1.5}H₁₆O: 212.120), 197 [M - Me] + (81), 169 [M - C₃H₇] + (22); 1 H NMR (400 MHz, CDCl₃): δ 9.25 (d, H-1), 7.53 (br d, H-2), 7.98 (br s, H-4), 7.56 (d, H-7), 7.89 (d, H-8), 3.82 (qq, H-11), 1.43 (d, H-12, H-13), 10.31 (s, H-14), 2.58 (br s, H-15); J [Hz]: 1,2=7,8=8; 11,12 = 11,13=7.

1β-Hydroxyambigu-4Z-ene (29). Colourless oil; IR $v_{max}^{\text{CCL}_4}$ cm $^{-1}$: 3620 (OH); MS m/z (rel. int.): 222.198 [M] $^+$ (5) (calc. for C_{1.5}H_{2.6}O: 222.198), 204 [M - H_{2.}O] $^+$ (10), 161 (10), 139 (55), 82 (100); ¹H NMR (400 MHz, C₆D₆, 57°): δ1.49 (ddd, H-2α), 1.23 (ddd, H-2β), 1.71 (m, H-3α), 2.84 (br dd, H-3β), 5.49 (dddq, H-5), 2.95 (br dd, H-6α), 1.76 (br dd, H-6β), 1.56 (m, H-9), 1.68 (m, H-10), 0.97 (m, H-10'), 1.59 (m, H-11), 0.91 (s, H-12), 1.23 (s, H-13), 1.79 (d, H-14), 0.82 (d, H-15), 0.71 (br s, OH); J [Hz]: 2α,2β = 14; 2α,3α = 2; 2α,3β = 7; 2β,3α = 12; 2β,3β = 2; 3α,3β = 14; 3,14 = 5,14 = 1.2; 5,6α = 10; 5,6β = 6.5; 6α,6β = 13; 8,9 = 8,9' = 9; 11,15 = 7; ¹³C NMR (C₆D₆, C-1-C-15): δ84.0, 27.1, 31.9, 139.7, 123.3, 41.2, 33.5, 52.3, 26.9, 37.3, 51.5, 30.6, 24.3, 24.5, 14.5; $[\alpha]_D^{24^\circ}$ - 45 (CHCl₃; c 0.89).

Methyl-8β-acetoxyguaia-4,11(13)-dien-12-oate (30). Colourless oil; IR $\nu_{\rm max}^{\rm CCl_4}$ cm $^{-1}$: 1740 (OAc), 1725 (CO $_2$ R); MS m/z (rel. int.): 306.183 [M] $^+$ (6) (calc. for C $_{18}$ H $_{26}$ O $_4$: 306.183), 274 [M - MeOH] $^+$ (1.3), 246 [M - HOAc] $^+$ (100), 231 (27), 187 (62), 105 (78), 91 (76); 1 H NMR (400 MHz, CDCl $_3$): δ 3.04 (br s, H-1), 1.93 and 1.44 (m, H-2), 2.23 (m, H-3), 2.69 (br dd, H-6), 2.23 (m, H-6'), 3.21 (br d, H-7), 5.15 (br dd, H-8), 2.05 (ddd, H-9), 1.74 (ddd, H-9'), 1.93 (m, H-10), 6.22 and 5.60 (br s, H-13), 0.87 (d, H-14), 1.58 (br s, H-15), 1.97 (s, OAc), 3.74 (s, OMe); 13 C NMR (CDCl $_3$, C-1-C-15): δ 39.2, 27.8, 37.3, 143.0, 133.4, 27.1, 52.0, 75.4, 39.3, 35.7, 119.2, 167.4, 124.7, 15.2, 14.2; OAc: 21.2, 170.3; OMe: 52.4; [α] $_2^{\rm Ca}$ $_2^{\rm Ca}$ (CHCl $_3$; c 1.74); Heating of 30 in dioxane-water with KOH for 4 hr afforded guaia-4,11(13),8 β ,12-olide [12].

11β-Acetoxy-8α-hydroxy-12α-senecioyloxy-ent-pimar-15-ene (31). Colourless gum; IR $\nu_{\rm max}^{\rm CCl_4}$ cm $^{-1}$: 3400 (OH), 1745, 1250 (OAc), 1720 (C=CCO₂R); MS m/z (rel. int.): 446.304 [M] $^+$ (2) (calc. for C₂₇H₄₂O₅: 446.303), 387 (M-OAc] $^+$ (2), 346 [M-RCO₂H] $^+$ (2), 286 [346-HOAc] $^+$ (8), 271 [286-Me] $^+$ (11), 83 [RCO] $^+$ (100); 1 H NMR (CDCl₃, 400 MHz): δ 1.45 (d, H-9), 5.76 (dd, H-11), 4.97 (d, H-12), 1.56 (d, H-14), 1.43 (d, H-14'), 5.67 (dd, H-15), 4.94 (dd, H-16t), 4.92 (dd, H-16c), 1.36 (d, H-17), 0.87 (d, H-18), 0.83 (d, H-19), 1.11 (d, H-20), 1.89 (d, OAc), 5.63 d, 2.14 d, 1.87 d (OSen); d [Hz]: 9,11 = 11.5; 11,12 = 10; 14,14′ = 14.5; 15,16t = 17.5; 15,16c = 11; OSen: 2',4′ = 2',5′ = 1.

19-Succinoyloxy-ent-pimara-7,14-diene (32). Isolated as its methyl ester 32a; colourless oil; IR $v_{\text{max}}^{\text{CCla}}$ cm⁻¹: 1735 (CO₂R), 3080, 1635, 920 (CH=CH₂); MS m/z (rel. int.): 402.277 [M]⁺ (8) (calc. for C₂₅H₃₈O₄: 402.277), 270 [M-RCO₂H]⁺ (100), 255 [270-Me]⁺ (9), 115 [RCO]⁺ (18); ¹H NMR (CDCl₃, 400 MHz): δ 1.83 (br d, H-1 α), 1.02 (dt, H-1 β), 1.45 (m, H-2), 0.96 (m, H-3 β), 1.75 (br d, H-3 α), 1.30 (m, H-5, H-11 β , H-12), 2.01 (br d, H-6 α), 1.90 (m, H-14 α , H-16 β), 5.34 (br d, H-7), 1.66 (m, H-9), 2.25 (dd, H-14 β), 5.67 (H-15), 4.95 (dd, H-16t), 4.97 (dd, H-16c), 0.96 (s, H-17), 0.95 (s, H-18), 4.39 (d, H-19), 3.98 (dd, H-19'), 0.78 (s, H-20); OCOR: 2.64 s, 3.69 s; J [Hz]: 1α , 1β = 1β , 2α = 3α , 3β ~ 13; 1β , 2β = 5; 3β , 19' = 1; 6α , 6β = 18; 6α , 7 = 5; 7, 14α = 2; 14α , 14β = 14; 15, 16t = 17; 15, 16c = 10.5; 16t, 16c = 1.5; 19, 19' = 10; $[\alpha]_{D}^{24^{\circ}}$ + 31 (CHCl₃; c 0.50).

Table 6. IR and MS data of compounds 1-24 (6 and 7 and 8 and 9 resp. as mixtures)

	IR $(v_{\text{max}} \text{ cm}^{-1})$		MS(m/z)
1*	3460, 1745, 1108	C ₂₀ H ₃₂ O ₄ (HP1, <i>R</i> , 9.7 min)	336.230 (11), 321 (50), 293 (43), 222 (27), 137 (55), 85 (70), 81 (100)
2	3560, 1750, 1725,	$C_{22}H_{34}O_{5}$	378.240 (5), 363 (35), 335 (26), 303 (8), 290 (15), 257 (24), 215 (26), 95 (60)
3	1250, 1215, 1090	(HP1, R_t 3.2 min)	81 (100)
3	3580, 1750, 1730, 1250, 1215, 1090	$C_{22}H_{34}O_5$	378.241 (10), 363 (40), 335 (30), 303 (10), 290 (12), 257 (20), 215 (14), 95 (100
4	3580, 1727, 1140,	(HP1, R_t 6.5 min)	81 (63)
•	1090	$C_{25}H_{38}O_5$ (HP2, R_t 6.9 min)	418.271 (6), 403 (20), 375 (15), 303 (4), 290 (14), 257 (12), 232 (10), 230 (10
5	3600, 1730, 1725,	$C_{25}H_{38}O_5$	83 (100), 81 (40), 55 (82) 418 271 (6), 402 (28), 275 (6), 202 (4), 285 (6), 275 (10), 257 (6), 82 (100)
J	1140, 1090, 1050	(HP2, R, 4.8 min)	418.271 (6), 403 (28), 375 (6), 303 (4), 285 (6), 275 (10), 257 (6), 83 (100), 81 (25), 55 (60)
6/7	1750, 1730, 1725,	$C_{30}H_{44}O_7$	516 (0.2), 501.286 (2) (calc. for C ₂₉ H ₄₁ O ₇ : 501.286), 385 (0.2), 285 (2), 257 (
0, 1	1270, 1240, 1170	(HP2, R, 7.2 min)	83 (100)
8/9	1740, 1730, 1720,	$C_{27}H_{40}O_6$	460.282 (5), 445 (15), 417 (4), 317 (9), 285 (6), 257 (7), 83 (100), 55 (68)
٥, ٥	1250, 1160, 1060	(HP2, R, 6.8 min)	400.202 (3), 443 (13), 417 (4), 317 (3), 283 (0), 237 (7), 83 (100), 33 (08)
10	1740, 1730, 1720,	$C_{27}H_{42}O_6$	462.298 (9), 447 (32), 419 (5), 317 (5), 285 (10), 257 (12), 57 (100)
	1250, 1130, 1060	(HP2, R, 7.5 min)	402,270 (7), 447 (32), 417 (3), 317 (3), 283 (10), 237 (12), 37 (100)
11	3600, 1730	$C_{25}H_{38}O_6$	434 (1), 419.244 (5). (calc. for C ₂₄ H ₃₅ O ₆ : 419.244), 323 (5), 255 (8), 205 (2
		(HP2, R, 3.4 min)	81 (100), 55 (60)
12	3600, 3400, 1725,	$C_{25}H_{38}O_6$	434 (0.5), 419.244 (4) (calc. for C ₂₄ H ₃₅ O ₆ : 419.244), 373 (3), 323 (5), 255 (7
	1150, 1090, 1050	$(T2 (3x), R_f 0.38)$	223 (12), 83 (100), 55 (78)
13	3600, 3450, 1740,	$C_{27}H_{40}O_7$	476.278 (5), 461 (10), 418 (6), 403 (20), 364 (8), 83 (100), 55 (80)
	1730, 1720	$(T2 (3x), R_c 0.33)$	(0), 03 (100), 33 (00)
14	3400, 1725	$C_{30}H_{44}O_{8}$	532 (1), 517.281 (5) (calc. for C ₂₉ H ₄₁ O ₈ : 517.280), 471 (3), 305 (3), 255 (10)
		$(T2 (3x), R_f 0.28)$	205 (22), 83 (100), 55 (40)
15†	3400, 1725	$C_{30}H_{44}O_{8}$	532 (0.5), 517.280 (3) (calc. for $C_{29}H_{41}O_8$: 517.280), 471 (2), 421 (2), 305 (
		(HP1, R, 11 min)	288 (2), 273 (2), 255 (4), 205 (12), 83 (100), 55 (20)
16	3480, 1750, 1726,	$C_{30}H_{44}O_{8}$	532 (0.5), 517.280 (8) (calc. for $C_{29}H_{41}O_8$: 517.280), 489 (2), 471 (4), 421 (3)
	1700	$(HP1, R_t 9.7 min)$	305 (3), 255 (10), 205 (25), 83 (100), 55 (50)
17	3500, 1750, 1730,	$C_{22}H_{32}O_{6}$	392 (0.5), 364.225 (13) (calc. for C ₃₁ H ₃₂ O ₅ : 364.225), 322 (4), 294 (8), 262 (20)
	1720	$(T2 (3x), R_f 0.59)$	234 (12), 216 (20), 83 (95), 55 (100)
8	3500, 1730, 1720	$C_{25}H_{36}O_{6}$	432. 252 (1), 414 (1), 404 (3), 363 (3), 334 (4), 279 (8), 262 (10), 234 (10), 216 (9)
		$(HP2, R_t 3.7 min)$	83 (100)
19	3600, 3500, 1740,	$C_{25}H_{38}O_{6}$	434 (0.2), 334.214 (5) (calc. for C ₂₀ H ₃₀ O ₄ : 334.214), 264 (6), 246 (15), 234 (8)
	1730, 1250, 1150	$(T1, R_f \ 0.39)$	231 (5), 220 (5), 203 (15), 83 (100), 55 (53)
20	3500-2600, 1730,	$C_{22}H_{34}O_{6}$	394.236 (4), 379 (40), 334 (5), 319 (9), 236 (12), 163 (25), 81 (100), 68 (75),
	1710, 1255	$(T4, R_f \ 0.44)$	55 (65)
1	3500-2600, 1735,	$C_{25}H_{38}O_{6}$	434.267 (5), 419 (14), 319 (6), 236 (18), 163 (20), 133 (28), 83 (100), 81 (40),
	1710	$(T4, R_f \ 0.33)$	55 (82)
22	3500-2600, 1730,	$C_{22}H_{34}O_{6}$	394.236 (6), 379 (16), 334 (7), 319 (6), 316 (4), 309 (15), 236 (40), 163 (45),
	1715, 1250	$(T4, R_f \ 0.43)$	81 (50), 57 (100)
23	3500–2600, 1735,	$C_{25}H_{38}O_6$	434.276 (4), 419 (4), 319 (3), 236 (8), 133 (36), 81 (66), 55 (100)
	1715	$(T4, R_f \ 0.30)$	
24	3600, 1730	$C_{25}H_{38}O_{6}$	434.267 (5), 419 (15), 334 (20), 316 (25), 301 (8), 261 (30), 243 (42), 189 (25)
		(HP2, R_i 5.8 min)	133 (25), 83 (100), 55 (75)

^{*}Mp 176°; †mp 187°.

 $\begin{array}{l} 12\alpha - Hydroxy - \mathrm{ent} - \mathrm{isokauren} - 19 - \mathrm{oic} \ \ \, \mathrm{acid} \ \, (33). \ \, \mathrm{Colourless} \ \, \mathrm{gum}; \\ \mathrm{IR} \ \, v_{\mathrm{max}}^{\mathrm{CHC1}}, \ \, \mathrm{cm}^{-1} \ \, 3500 - 2600, \ \, 1700 \ \, (\mathrm{CO}_2\mathrm{H}); \ \, \mathrm{MS} \ \, \mathit{m/z} \ \, \mathrm{(rel.\ int.)}; \\ 318.219 \ \, [\mathrm{M}]^+ \ \, (24) \ \, (\mathrm{calc.\ \, for} \ \, \mathrm{C}_{20}\mathrm{H}_{30}\mathrm{O}_{3}; \ \, 318.219), \ \, 300 \ \, [\mathrm{M} - \mathrm{H}_2\mathrm{O}]^+ \ \, (6), \ \, 107 \\ \mathrm{[C}_8\mathrm{H}_{11}]^+ \ \, (100); \ \, ^{1}\mathrm{H} \ \, \mathrm{NMR} \ \, (400 \ \, \mathrm{MHz}, \mathrm{C}_6\mathrm{D}_6); \ \, \delta \, 2.10 \ \, (\mathit{br} \ \, \mathit{d}, \mathrm{H} - 1\alpha), \\ 0.89 \ \, (\mathit{dt}, \mathrm{H} - 1\beta), \ \, 2.34 \ \, (\mathit{br}, \mathrm{H} - 2\alpha), \ \, 1.60 \ \, (\mathit{br} \ \, \mathit{d}, \mathrm{H} - 2\beta), \ \, 2.41 \ \, (\mathit{br} \ \, \mathit{d}, \mathrm{H} - 3\alpha), \\ 0.98 \ \, (\mathit{dt}, \mathrm{H} - 3\beta), \ \, 2.00 \ \, (\mathit{br} \ \, \mathit{d}, \mathrm{H} - 5), \ \, 2.18 \ \, (\mathit{dg}, \mathrm{H} - 6\alpha), \ \, 1.05 \ \, (\mathit{dd}, \mathrm{H} - 6\beta), \\ 1.80 \ \, (\mathit{br} \ \, \mathit{d}, \mathrm{H} - 7\alpha), \ \, 1.56 \ \, (\mathit{ddd}, \mathrm{H} - 7\beta), \ \, 1.14 \ \, (\mathit{d}, \mathrm{H} - 9), \ \, 1.94 \ \, (\mathit{dt}, \mathrm{H} - 11), \\ 1.77 \ \, (\mathit{br} \ \, \mathit{d}, \mathrm{H} - 11\beta), \ \, 4.56 \ \, (\mathit{dd}, \mathrm{H} - 12), \ \, 1.54 \ \, (\mathit{br} \ \, \mathit{t}, \mathrm{H} - 13), \ \, 1.46 \ \, (\mathit{dd}, \mathrm{H} - 14\alpha), \\ 2.76 \ \, (\mathit{br} \ \, \mathit{d}, \mathrm{H} - 14\beta), \ \, 5.09 \ \, (\mathit{q}, \mathrm{H} - 15), \ \, 1.73 \ \, (\mathit{br} \ \, \mathit{s}, \mathrm{H} - 17), \ \, 1.27 \ \, (\mathit{s}, \mathrm{H} - 18), \\ 1.47 \ \, (\mathit{s}, \mathrm{H} - 20); \ \, \mathit{J} \ \, [\mathrm{Hz}]: \ \, 1\alpha, 1\beta = 1\beta, 2\alpha = 2\alpha, 2\beta = 3\alpha, 3\beta = 2\alpha, 3\beta = 13; \\ 1\beta, 2\beta = 2\beta, 3\beta \sim 4; \ \, 5.6\alpha = 12; \ \, 6\alpha, 6\beta = 6\alpha, 7\beta = 13; \ \, 6\alpha, 7\alpha = 2; \\ 6\beta, 7\beta = 2; \ \, 9.11\alpha = 9; \ \, 11\alpha, 11\beta = 10; \ \, 11\beta, 12 = 5; \ \, 12, 13 = 13, 14\alpha = 3; \\ 14\alpha, 14\beta = 12; \ \, 15, 17 = 1; \ \, ^{13}\mathrm{C} \ \, \mathrm{NMR} \ \, (\mathrm{C}_6\mathrm{D}_6, 67.8 \ \, \mathrm{MHz}, \mathrm{C} - 1 - \mathrm{C} - 20); \\ \end{array}$

δ 39.5, 19.6, 38.4, 44.1, 56.7, 21.1, 37.2, 48.7, 51.3, 33.7, 29.4, 66.7, 49.0, 41.0, 138.1, 141.6, 15.7, 28.4, 182.2, 15.7; $[\alpha]_D^{24^*}$ –62 (CHCl₃; c 1.03).

5-[5-Hydroxypenta-1,3-diynyl]-2-[but-3-en-1-ynyl]-thiophene (37). Colourless gum; IR $v_{\rm max}^{\rm CCL}$ cm $^{-1}$: 3600 (OH), 2200 (C \equiv C); UV $\lambda_{\rm max}$ (Et₂O): 350, 330 nm; MS m/z (rel. int.): 212.030 [M] $^+$ (100) (calc. for C₁₃H₈OS: 212.030); 1 H NMR (CDCl₃, 400 MHz): 5.76 (dd, H-1t), 5.60 (dd, H-1c), 6.00 (dd, H-2), 7.03 (d, H-6), 7.17 (d, H-7), 4.44 (s, H-13).

REFERENCES

 Merxmüller, H., Leins, P. and Roessler, H. (1977) in The Biology and Chemistry of the Compositae (Heywood, V. H.,

- Harborne, J. B. and Turner, B. L., eds), p. 592. Academic Press, New York.
- Jakupovic, J., Lehmann, L., Bohlmann, F., King, R. M. and Robinson, H. (1988) Phytochemistry 27, 3831.
- 3. Lehmann, L., Jakupovic, J., King, R. M. and Robinson, H. (1988) *Phytochemistry* 27, 2994.
- Jakupovic, J., Schuster, A., Bohlmann, F., King, R. M. and Robinson, H. (1988) Phytochemistry 27, 3181.
- Zdero, C., Bohlmann, F., King, R. M. and Robinson, H. (1989) Phytochemistry 28, 517.
- 6. Jakupovic, J., Pathak, V. P., Bohlmann, F., King, R. M. and Robinson, H. (1987) *Phytochemistry* 26, 803.
- 7. Zdero, C., Bohlmann, F., King, R. M. and Robinson, H. (1987) Phytochemistry 26, 1759.
- 8. Zdero, C., Bohlmann, F., Haegi, L. and King, R. M. (1987) Liebigs Ann. Chem. 665.

- 9. Zdero, C., Bohlmann, F., Haegi, L. and King, R. M. (1988) Phytochemistry 27, 685.
- 10. Bohlmann, F. and Zdero, C. (1973) Chem. Ber. 106, 1337.
- 11. Bohlmann, F. and Zdero, C. (1975) Chem. Ber. 108, 362.
- 12. Zdero, C., Bohlmann, F., King, R. M. and Robinson, H. (1988) Phytochemistry 27, 2835.
- Matthes, H. W. D., Luu, B. and Ourisson, G. (1982) Tetrahedron 38, 3129.
- Bohlmann, F., Arndt, C., Kleine, K. M. and Bornowski, H. (1965) Chem. Ber. 98, 155.
- Bohlmann, F., Grenz, M., Wotschokowsky, M. and Berger, E. (1967) Chem. Ber. 100, 2518.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1984) Phytochemistry 23, 1979.